

3-Benzylidene-1-(2,6-dichlorophenyl)indolin-2-one

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Key indicators

Single-crystal X-ray study

$T = 293\text{ K}$

Mean $\sigma(\text{C}-\text{C}) = 0.004\text{ \AA}$

R factor = 0.042

wR factor = 0.107

Data-to-parameter ratio = 13.6

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The phenyl and dichlorophenyl rings of the title molecule, $\text{C}_{21}\text{H}_{13}\text{Cl}_2\text{NO}$, are oriented at angles of $39.3(1)^\circ$ and $77.8(1)^\circ$, respectively, with respect to the central indoline ring. The crystal structure is stabilized by van der Waals interactions.

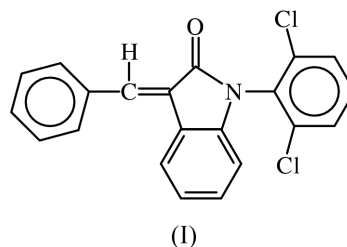
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Comment

Indole derivatives, widely distributed in living cells as tryptophan metabolites, have important biological functions. They are found to possess antihypertensive (Merk, 1971, 1974), anti-inflammatory (Rodriguez *et al.*, 1985) and antimalarial (El-Sayed *et al.*, 1986) activities. The structure determination of the title compound, (I), was undertaken to study the effect of the substituent groups on the conformation of the indoline ring and the molecular stereochemistry.



The phenyl ring is oriented at an angle of $39.3(1)^\circ$ with respect to the indoline ring. The dihedral angle between the dichlorophenyl ring and the indoline ring is $77.8(1)^\circ$. The substituent groups at N1 and C2 of indoline do not disturb the planarity of that ring. The Csp^2-Cl bond lengths [$\text{C14}-\text{Cl1} = 1.725(3)\text{ \AA}$ and $\text{C10}-\text{Cl2} = 1.728(3)\text{ \AA}$] are in good agreement with the literature values (Allen *et al.*, 1987). The exocyclic angle $\text{C2}-\text{C15}-\text{C16}$ of $129.5(2)^\circ$ deviates significantly from the normal value of 120° . This may be due to the repulsion between atoms H3 and H17 (2.31 \AA). It is of interest to note that the shortest $\text{Cl}\cdots\text{Cl}$ intermolecular distance is $3.409(2)\text{ \AA}$, slightly smaller than the sum of the van der Waals radii of the corresponding atoms. The crystal structure is stabilized by van der Waals interactions.

Experimental

A mixture of salicylaldehyde (0.01 mol), dichlorofenic acid (0.01 mol) and triethylamine (2 ml) was refluxed on an oil bath for 5 h. On cooling, a crystalline solid separated out, was filtered off and then washed with chilled methanol and recrystallized from ethanol (yield: 75%; m.p. $390-391\text{ K}$).

